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concepts through real-
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branch of chemistry
that studies the
structure, properties
and reactions of
organic compounds,
which contain carbon in*

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covalent bonding. Study of structure determines their chemical composition and formula. Study of properties includes physical and chemical properties, and evaluation of chemical reactivity to understand their behavior.

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lead compounds.

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Chemistry (IUPAC). It is
published in the
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Organic Chemistry
(informally called the
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every possible organic
compound should have
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*General Chemistry: The
Essential Concepts
(sixth and tenth
edition) is written by
Raymond Chang
(Williams College) and
Jason Overby (The
College of Charleston)*

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and published by

McGraw-Hill in 2011.

This book is fully colored and very helpful to understand the basics of general chemistry. This book is also known as Chang General Chemistry.

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An organic compound is defined as any

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compound whose molecules contain carbon and hydrogen (also known as " hydrocarbons") or compound that is the derivative of it. The branch of science which deals with the scientific study of structure, properties and reactions of hydrocarbons and their derivatives is known as organic chemistry.

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Many more linear alkanes can be formed by adding one additional carbon to the end of a chain of carbons. Ethane is the shortest chain with two carbons but DNA is known to have carbon chains containing

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Functional groups are

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specific groupings of atoms within molecules that have their own characteristic properties, regardless of the other atoms present in a molecule. Common examples are alcohols, amines, carboxylic acids, ketones, and ethers. In a typical sophomore organic chemistry course, there's about 14 functional groups

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*that are key, with
another group of 8 that
make appearances
from ...*

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introductory university
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